Chapter 5: How Ecosystems Work Section 2, The Cycling of Materials

The Carbon Cycle

•	The	is the movement of carbon from the nonliving environment into living things
	and back	
•	Carbon is the essential component of	which make up all
	organisms.	
The Ca	rbon Cycle	
•	Carbon exists in	.
•	Producers convert	in the atmosphere into carbohydrates during
	photosynthesis.	Atmospheric carbon dioxide, CO ₂
•	Consumers obtain carbon from the	
	carbohydrates in the producers they	RESPIRATION COMBUSTION
	eat.	RESPIRATION PHOTOSYNTHESIS (human and natural)
The Ca	rbon Cycle	Natural EROSION DECOMPOSITION
•	During cellular respiration, some of	Gas CO ₂ dissolved Plant and animal remains Coal
	the carbon is released back into the	Limestone - Marine plankton remains Natural gas
	atmosphere as	51
		·
•		, forming one of the largest
The Co	when Cools	on Earth.
ine Ca	rbon Cycle	
•	Carbon stored in the bodies of organisms	s as, may
	be released into the soil or air when the	organisms dies.
•	These molecules may form deposits of _	, which are
	known as	·

Fossil fuels store carbon left over from bodies of organisms that dies millions of years ago.

How Humans Affect the Carbon Cycle

- Humans burn fossil fuels, releasing carbon into the atmosphere.
- The carbon returns to the atmosphere as _______

How Humans Affect the Carbon Cycle

- Increased levels of carbon dioxide may contribute to _______.
- Global warming is an _____ of the Earth.

The Nitrogen Cycle

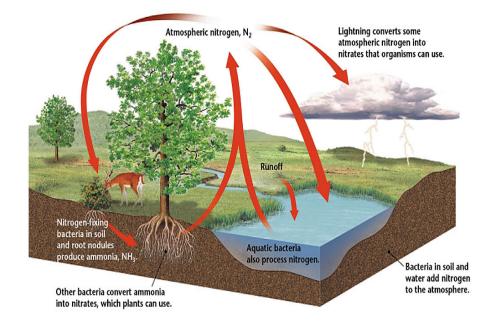
- The ______ is the process in which nitrogen circulates among the air, soil, water, plants, and animals in an ecosystem.
- All organisms need nitrogen to ______, which are used to build new cells.
- Nitrogen makes up ______ percent of the gases in the atmosphere.

The Nitrogen Cycle

- Nitrogen must be ______, before organisms can use it.
- Only a few species of bacteria can fix atmospheric nitrogen into chemical compounds that can be used by other organisms.
- These bacteria are known as ______ bacteria.

The Nitrogen Cycle

- are bacteria that convert atmospheric nitrogen into ammonia.
- The bacteria use sugar provided by the legumes to produce nitrogen-containing compounds such as
- Excess nitrogen fixed by the bacteria is released into the soil.



Decomposers and the Nitrogen Cycle

•	Nitrogen stored within the bodies of living things is returned to the nitrogen cycle once those organisms die.				
•	break down decaying plants and animals, as well as plant and animal				
	wastes.				
•	After decomposers return nitrogen to the soil, bacteria transform a small amount of the nitrogen into				
	, which then returns to the	atmosphere to complete the nitrogen cycle.			
The Pho	Phosphorus Cycle				
•	Phos	phate ng			
	element that is part of many molecules that make up				
	the cells of living organisms.	osphate			
•	Plants get the phosphorus they need from	Runoff			
	, while animals get	Phosphate in water			
	their phosphorus by	Decomposition of plants and animals Phosphate in soil			
that have eaten plants.					
•	• The is the cyclic	movement of phosphorus in different chemical forms			
	from the environment to organisms and then back to the environment.				
The Pho	Phosphorus Cycle				
•	Phosphorus may enter soil and water when rocks erode.				
•	Small amounts of phosphorus dissolve as, which moves into the soil.				
•	Plants absorb phosphates in the soil through their roots.				
•	Some phosphorus washes off the land and ends up in the ocean.				
•	 Because many phosphate salts are not soluble in water, they sink to the bottom and 				

•	, which people use to stimulate and maximize plant growth, contain both nitrogen and			
	phosphorus.			
•	Excessive amounts of fertilizer can enter terrestrial and aquatic ecosystems through			
	·			
•	Excess nitrogen and phosphorus can cause			
•	Excess algae can deplete an aquatic ecosystem of important nutrients such as, o			
	which fish and other aquatic organisms depend.			
Acid P	recipitation			
•	When fuel is burned, large amounts of is release into the atmosphere.			
•	In the air, nitric oxide can combine with oxygen and water vapor to form			
•	Dissolved in rain or snow, the nitric acid falls as			

Fertilizers and the Nitrogen and Phosphorus Cycles